East English Village Preparatory Academy

Course Syllabus-11th Grade AP Chemistry Academic Year 2021_2022

Course Information General Chemistry Marking Period: 1, (8-9.10am) Class Location: Room A110 Time: Period 1

Teacher Information

Name: Dr. Anura Hewagama Phone: Google Voice: (734) 623-9052 Office Location: Room A110 Office Hours: Mon –Friday 3.30-4.00pm or By appointment. E-Mail: <u>anura.hewagama@detroitk12.org</u> Website: <u>https://ahewagama.weebly.com/</u>

DPSCD MISSION AND VISION:

We educate and empower every student, in every community, every day, to build a stronger Detroit. All students will have the knowledge, skills and confidence necessary to thrive in our city, our nation, our world.

COURSE DESCRIPTION:

The course is designed for 11th-12th graders who are planning College level education. Almost all the Science majors and some non-science level College programs require you to complete a course in General Chemistry. It is designed to provide a practical approach to the chemistry in our world that we are surrounded by every day. The curriculum is aligned with College board (https://www.collegeboard.org/) AP CHEMISTRY curriculum.

COURSE OBJECTIVES:

The students will be able to:

- 1. Describe the nature of scientific investigation
- 2. Use Mathematical Manipulations (Quantitative data) in Scientific Observation
- Demonstrate knowledge in concepts relevant to high school chemistry/first Year College curriculum standards
- 4. Be knowledgeable in basic laboratory techniques used in a chemistry laboratory
- 5. To understand how, science, technology and society relate to each other.
- 6. Prepare well for the State of Michigan Merit Exam

UNITS

Students who successfully complete Chemistry will be competent in the following areas:

- Explain the structure of matter.
- Explain how changes in matter are accompanied by changes in energy.
- Describe the relationship between temperature, pressure, volume, and the amount of gas.
- Predict products of simple reactions and balance chemical equations.

- Identify signs that indicate a chemical reaction has taken place.
- Stoichiometric Calculations for chemical reactions.
- Apply the mole concept in stoichiometric calculations including those involving limiting reactants and percent yield.

Class Requirements

- 1. In order to be a member of this class, you and your parent/guardian will be asked to read and sign a contract. This document contains classroom grading policies as well as expectations for the course. If you do not consistently follow instructions and, abide by the classroom policies you will not be able to achieve the expectations of this course. Therefore, it is crucial that you understand and sign this document in order to participate in this class.
- 2. All students must have a account on Microsoft Teams and on Schoology.
- 3. Your course material, assignments and the gradebook will be maintained in weekly folders in Schoology.
- 4. Furthermore, it is required that you maintain a notebook part of class participation. Credits will be given for well-maintained grade class notebooks for each grading period. Each day, "bell work(Do-Now" will be assigned and completed at the beginning of class. I will periodically be checking to see that you 1) have your notebook OR binder in your possession and 2) that you are regularly completing the "bell work" assignments.

Classroom Policies

All students are required to follow classroom policies. You earn daily 10 compliance policies for not violating any of the following policies.

- 1. Maintain quiet learning environment.
- 2. Do not disrupt the classroom routine,
- 3. Do not use cell phones without teacher permission
- 4. Do not bring food or drinks to the classroom. Do not eat or drink in the classroom,

COURSE MATERIAL -IN SCHOOLOGY

STUDENT EVALUATION

The grading system for the AP Chemistry Course at East English Village Preparatory Academy is as follows:

	Commented [AH1]:
40%	
20%	
ur grade per semester. In addition, your	
vior and the	
10%	
	20% Ir grade per semester. In addition, your fior and the

Attendance Policy

Regular and prompt class attendance is an essential part of the educational experience. EEVPA expects students to exercise good judgment regarding attendance and absences. Students will accept full responsibility for ensuring their work does not suffer because of absences. All students are expected to attend every scheduled class on time. Exceptions may be made for illness and valid emergencies.

Classroom Expectations

- 1. You are expected to bring all the writing supplies every day to the classroom.
- 2. During the online learning- make your best effort to participate in the lesson.
- 3. RESPECT others' work, ideas.
- 4. Cooperate with others.
- 5. Make sure you make a full use of the class hour. Any distraction is not accepted.
- 6. Enter and leave the room online session in a respectful manner. Make sure to sign in using MS Teams Chat option as soon as you enter the classroom.
- 7. Be safe! Use your time wisely, avoid distractions.

Plagiarism, Cheating, and Academic Integrity

Plagiarism is the practice of copying words, sentences, images, or ideas for use in written or oral assessments without giving proper credit to the source. Cheating is defined as the giving or receiving of illegal help on anything that has been determined by the teacher to be an individual effort. Both are considered serious offenses and will significantly affect your course grade. Please refer to the Student Code of Conduct booklet for additional information.

Methodology

A combination of lecture, class discussion, presentations, videos, cooperative learning, and problembased learning will be used in this course. Grades will be determined by the satisfactory and timely completion of assignments. The grade of each assignment is based on the prerequisite given for each assignment. Below is an overview of topic/ units and major assessments/assignments for this course. Please note dates/timeframes are subject to change and are an estimate.

Unit	Course Activities	Assessments	Month/ Timeframe
Introduction to Chemistry/Properties of Matter	Worksheets, guided reading, lecture, laboratory	Unit Test	Sept/October
Atomic Theory	Worksheets, candy isotope, guided reading, History of the Atom research, lecture	Unit Test	October 3 ^{rd,} Week
Behavior of the Atom	Electron map, worksheets, guided reading, lecture	Unit Test	October 4 th and Nov 1st Weeks
Periodic Trends	Trends WebQuest, lecture, worksheets, guided reading	Unit Test	Nov. 2 nd week to December 3 rd week
Chemical Reactions	Laboratory, lecture, worksheets, guided reading	Unit Test	Dec. & Jan./2 nd MP

General Information

Bathroom Pass Policy:

Make sure to regular requests for bathroom passes. Do not take more time than necessary to use bathrooms. You are expected return to the class in 5 minutes after receiving a bathroom pass. Violations of the bathroom pass policies will result in log entries and denial of your future requests. No passes are given during the first 15 and last 15 minutes of the class hour.

Textbook:

For general High School level: I use CK.12 online textbook for the class (https://flexbooks.ck12.org/cbook/ck-12-chemistry-flexbook-2.0/). Advanced classwork will be completed along with Khan Academy AP Chemistry Course work(https://www.khanacademy.org/science/ap-chemistry-beta)

The textbook used in this course is titled *Modern Chemistry* by Holt, Rinehart, and Winston. You will be given access to an online version of the textbook. Hard copies of the textbook will be stored in the classroom and used as needed. If you do not have access to the internet at home, please let me know and you will be allowed to sign out a hard copy of the textbook. If at any time you feel that having a hard copy of the textbook would better suit you, you may check with me and sign one out. Just remember that while the book is in your possession, it is your responsibility. Any damages to the book that occur while it is in your possession will need repaired and therefore will result in a monetary fine. The cost to replace a lost book is \$75.00. Any fees that remain unpaid can result in the withholding of your report card at the end of a marking period.

Good, better, best, will never do you rest; until your good is better, until your better is best. Let's do our best this semester and also have some fun!